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*Phytochemistry*, Vol. 26, No. 9, pp. 2639–2641, 1987.  
 Printed in Great Britain.

0031-9422/87 \$3.00 + 0.00  
 Pergamon Journals Ltd.

## LABDANE DITERPENES FROM *BRICKELLIA GLOMERATA*\*

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(Received 16 December 1986)

**Key Word Index**—*Brickellia glomerata*; Compositae; Eupatorieae; labdane type diterpenes; *enantio*-oliveric acid; 12-oxolambertianic acid; demethyl pinusolide; lambertianic acid.

**Abstract**—The leaves of *B. glomerata* afforded three new labdane diterpenes. Their structures and stereochemistry were established by spectroscopic methods and chemical transformations.

### INTRODUCTION

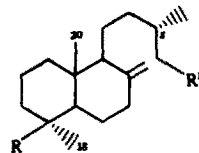
In continuation of our chemical studies of members of the genus *Brickellia* (Eupatorieae) [1–4], we have undertaken the study of *B. glomerata* and have isolated in addition to the known lambertianic acid, three new labdane type diterpenes whose structures were established as *enantio*-oliveric acid (1a), 12-oxolambertianic acid (2b) and demethyl pinusolide (3a).

### RESULTS AND DISCUSSION

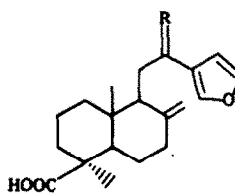
Compound 1a was the major constituent of the leaves of *B. glomerata*,  $C_{20}H_{32}O_4$ ,  $[M]^+$  at  $m/z$  336, was isolated as a gum. The presence in 1a of carboxylic acid group(s) were shown by IR absorptions at 3500–2380 and 1695  $cm^{-1}$ , and by the formation of the dimethyl ester (1b) upon treatment with excess of diazomethane. It also contained an exocyclic methylene group (IR absorptions at 1640 and 890  $cm^{-1}$ ). The  $^1H$  NMR spectrum of 1a showed signals due to two tertiary methyl groups ( $\delta$ 0.55 and 1.27) and one secondary methyl group ( $\delta$ 0.95,  $d$ ,  $J = 6.5$  Hz). The presence of the exocyclic methylene group was confirmed by absorptions at  $\delta$ 4.47 and 4.80 (*br s*, 1H each).

The hindered nature of one of the carboxylic groups in 1a followed from the formation of the mono ester 1c, obtained by Fischer esterification, while alkaline hydro-

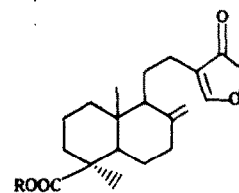
lysis of 1b gave the isomeric mono-ester 1d. Additional proof for the axial configuration of the carboxyl group at C-4 was provided by reduction of 1a with lithium aluminium hydride to the diol 1e. The  $^1H$  NMR spectrum of 1e, showed two doublets at  $\delta$ 3.36 and 3.75 (AB system) characteristic of an axial hydroxy methylene group at C-4 [5–6].



- 1a R = R' = COOH  
 1b R = R' = COOMe  
 1c R = COOH, R' = COOMe  
 1d R = COOMe, R' = COOH  
 1e R = R' = CH<sub>2</sub>OH



- 2a R = H<sub>2</sub>  
 2b R = O



- 3a R = H  
 3b R = Me

\*Contribution No. 844 from Instituto de Química, UNAM.

The stereochemistry at C-13 and the absolute configuration of **1a** were established by comparison of the  $^1\text{H}$  NMR data of **1b** with those reported for dimethyl oliverate [7], which were identical, but the optical rotation had opposite sign. Hence **1a** must be (13*S*)-labd-8(17)-en-15,19-dioic acid (*enantio*-oliveric acid).

12-Oxo-lambertianic acid (**2b**) was a gum,  $\text{C}_{20}\text{H}_{26}\text{O}_4$  ( $[\text{M}]^+$  at  $m/z$  330), which exhibited bands in the IR spectrum at 3700–2400 and  $1690\text{ cm}^{-1}$  corresponding to a carboxylic acid group. The structure of **2b** clearly followed from the  $^1\text{H}$  NMR spectrum which was very similar to that of lambertianic acid (**2a**) [8], differing only with respect to the furan ring signals which were downfield shifted, suggesting a  $\beta$ -ketofuran grouping [9]. This assumption was confirmed by the presence in the electron impact mass spectrum of the base peak at  $m/z$  95 [9]. The axial orientation of the carboxyl group at C-4 was deduced from the typical position of the C-10 methyl signal [8, 10].

A third isolated compound (**3a**) was a gum with  $[\text{M}]^+$  at  $m/z$  332 indicating the formula  $\text{C}_{20}\text{H}_{28}\text{O}_4$ . The IR spectrum showed the presence of a carboxylic acid ( $3450$ – $2400$  and  $1693\text{ cm}^{-1}$ ) and a  $\gamma$ -lactone ( $1753\text{ cm}^{-1}$ ). The structure of **3a** followed from the  $^1\text{H}$  NMR spectrum which showed a great resemblance to the data described for nivenolide [11].

The stereochemistry of the carboxyl group at C-4 and the absolute configuration of **3a**, was established by conversion to the methyl ester **3b**, whose  $^1\text{H}$  NMR and optical rotation data were identical to those previously published for pinusolide [12].

#### EXPERIMENTAL

*Brickellia glomerata* Fernald, was collected near Amacuzac, on the toll road Amacuzac–Iguala (59.5 km), state of Morelos, in December 1983. A voucher specimen has been deposited in the Herbarium of the University of Texas at Austin, Texas. Dried leaves (940 g) were extracted with  $\text{CH}_2\text{Cl}_2$  and the extract was evaporated *in vacuo* to give a gummy residue (388 g). A 10.5 g sample was chromatographed over 110 g silica gel packed in  $\text{CH}_2\text{Cl}_2$ , 125 ml fractions being collected as follows: 1–9 ( $\text{CH}_2\text{Cl}_2$ ), 10–18 ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$ , 19:1), 19–22 ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$ , 9:1).

Fractions 4 and 5 on standing afforded 300 mg of **2a** as a colourless solid, mp  $124^\circ$  (lit.  $126$ – $127^\circ$  [8]),  $[\alpha]_D^{25} = +45.5^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.47) (lit.  $[\alpha]_D^{25} = +55^\circ$  (EtOH) [8]). Fractions 10 and 11 (1.72 g) were rechromatographed in 100 g of silica gel using  $\text{CH}_2\text{Cl}_2$  and mixtures of  $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$  as eluants. Initial fractions were further purified by prep. TLC ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$  19:1,  $\times 2$ ) gave **2a** [8], **2b** and **3a**.

12-Oxo lambertianic acid (**2b**). A gum, IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3700–2400, 1690, 890.  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.70 (3H, s, H-20), 1.27 (3H, s, H-18), 4.36 (1H, *br s*, H-17), 4.72 (1H, *br s*, H-17'), 6.76 (1H, *br s*, H-14), 7.41 (1H, *t*,  $J = 1.5\text{ Hz}$ , H-15), 8.05 (1H, *br s*, H-16). EIMS (probe) 70 eV  $m/z$  (rel. int.): 330  $[\text{M}]^+$  (9), 220 ( $\text{M} - \text{Me} - 95$ ) $^+$  (9), 121 ( $\text{C}_9\text{H}_{13}$ ) $^+$  (18), 95 ( $+ \text{O} = \text{C} - \text{C}_4\text{H}_5\text{O}$ ) (100).

Demethyl pinusolide (**3a**). Colourless gum; IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3450–2400, 1754, 1694, 895.  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.60 (3H, s, H-20), 1.25 (3H, s, H-18), 4.55 (1H, *br s*, H-17), 4.85 (1H, *br s*, H-17'), 4.75 (2H, *br s*, H-15), 7.05 (1H, *br s*, H-14). EIMS (probe) 70 eV  $m/z$  (rel. int.): 332  $[\text{M}]^+$  (5), 314 ( $\text{M} - \text{H}_2\text{O}$ ) $^+$  (7), 286 ( $\text{M} - \text{CH}_2\text{O}_2$ ) $^+$  (19), 271 ( $\text{M} - \text{CH}_2\text{O}_2 - \text{Me}$ ) $^+$  (17), 121 ( $\text{C}_9\text{H}_{13}$ ) $^+$  (51), 91 (62), 41 (100). Compound **3a** on esterification with  $\text{CH}_2\text{N}_2$  afforded **3b**. The  $^1\text{H}$  NMR spectrum was identical to that previously published for pinusolide (**3b**) [12].

Fractions 13–16 exhibited one major spot and were combined to give 1.3 g of crude **1a**. A 100 mg sample was purified by prep. TLC using  $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$  (19:1) as eluant and afforded 60 mg of *enantio*-oliveric acid (**1a**) as a colourless gum.  $[\alpha]_D^{25} + 30.5^\circ$  (EtOH;  $c$  0.59); IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3500–2380, 1695, 1640, 890;  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.55 (3H, s, H-20), 0.95 (3H, *d*,  $J = 6.5\text{ Hz}$ , H-16), 1.27 (3H, s, H-18), 4.47 (1H, *br s*, H-17), 4.80 (1H, *br s*, H-17'). EIMS, 70 eV,  $m/z$  (rel. int.): 336  $[\text{M}]^+$  (3), 318 ( $\text{M} - \text{H}_2\text{O}$ ) $^+$  (8), 300 ( $\text{M} - 2\text{H}_2\text{O}$ ) $^+$  (5), 290 ( $\text{M} - \text{HCOOH}$ ) $^+$  (22), 121 ( $\text{C}_9\text{H}_{13}$ ) $^+$  (100), 55 ( $\text{C}_4\text{H}_7$ ) $^+$  (55).

*enantio*-Dimethyl oliverate (**1b**). Esterification of crude **1a** (616 mg) with excess of  $\text{CH}_2\text{N}_2$  provided 300 mg of the ester **1b** after TLC purification ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$ , 19:1). Colourless gum.  $[\alpha]_D^{25} + 49.5^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.58); IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 1720, 890;  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.50 (3H, s, H-20), 0.94 (3H, *d*,  $J = 6.0\text{ Hz}$ , H-16), 1.18 (3H, s, H-18), 3.60 (3H, s, O–Me), 3.65 (3H, s, O–Me), 4.46 (1H, *br s*, H-17), 4.81 (1H, *br s*, H-17').

*enantio*-Oliveric acid, 15-monomethyl ester (**1c**). Compound **1a** (291 mg) in MeOH and a drop of conc  $\text{H}_2\text{SO}_4$  was refluxed for 1.3 hr and worked-up as usual to give, after TLC purification, 273 mg of **1c**. Colourless oil. IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3500–2500, 1730, 1698, 1650, 892.  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.60 (3H, s, H-20), 1.24 (3H, s, H-18), 0.93 (3H, *d*,  $J = 6.0\text{ Hz}$ , H-16), 3.65 (3H, s, O–Me), 4.45 (1H, *br s*, H-17), 4.80 (1H, *br s*, H-17').

*enantio*-Oliveric acid, 19-monomethyl ester (**1d**). To a soln of 247 mg of **1b** in 20 ml MeOH, 127 mg NaOH was added. The reaction mixture was refluxed for 2 hr and worked-up as usual to give, after TLC purification ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$ , 19:1) 196 mg **1d** as a colourless oil. IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3500–2500, 1712, 890. EIMS 70 eV,  $m/z$  (rel. int.): 350  $[\text{M}]^+$  (6), 290 ( $\text{M} - 60$ ) $^+$  (31), 121 ( $\text{C}_9\text{H}_{13}$ ) $^+$  (100).  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.50 (3H, s, H-20), 0.97 (3H, *d*,  $J = 6.0\text{ Hz}$ , H-16), 1.18 (3H, s, H-18), 3.60 (3H, s, O–Me), 4.46 (1H, *br s*, H-17), 4.81 (1H, *br s*, H-17').

(13*S*)-Labd-8(17)-en-15,19 diol (**1e**). To a soln of **1d** (189 mg) in 30 ml dry  $\text{Et}_2\text{O}$  was added  $\text{LiAlH}_4$  (391 mg) in small amounts, the reaction being monitored by TLC. After 1 hr the reaction was worked-up and the residue purified by TLC ( $\text{CH}_2\text{Cl}_2$ – $\text{Me}_2\text{CO}$ , 19:1), yielding 65 mg of diol **1e** as colourless crystals, mp  $98$ – $100^\circ$ ,  $[\alpha]_D^{25} + 35^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.439), IR  $\nu_{\text{max}}^{\text{CHCl}_3}\text{ cm}^{-1}$ : 3615, 1645, 890; EIMS 70 eV  $m/z$  (rel. int.): 308  $[\text{M}]^+$  (5), 290 ( $\text{M} - \text{H}_2\text{O}$ ) $^+$ , 277 ( $\text{M} - \text{CH}_2\text{OH}$ ) $^+$  (80), 121 ( $\text{C}_9\text{H}_{13}$ ) $^+$  (60), 95 (100), 55 (92).  $^1\text{H}$  NMR (80 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.64 (3H, s, H-20), 0.90 (3H, *d*,  $J = 6.0\text{ Hz}$ , H-16), 0.97 (3H, s, H-18), 3.65 (2H, *t*,  $J = 6.5\text{ Hz}$ , H-15), 3.36 (1H, *d*,  $J = 11.5\text{ Hz}$ , H-19), 3.75 (1H, *d*,  $J = 11.5\text{ Hz}$ , H-19'), 4.47 (1H, *br s*, H-17), 4.77 (1H, *br s*, H-17').

**Acknowledgements**—We thank Prof. B. L. Turner, University of Texas at Austin, for identification of the plant material and Messrs J. Cárdenas, H. Bojórquez, L. Velasco and R. Villena for  $^1\text{H}$  NMR, IR and EM spectra and optical rotation data.

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*Phytochemistry*, Vol. 26, No. 9, pp. 2641–2643, 1987.  
Printed in Great Britain.

0031-9422/87 \$3.00 + 0.00  
Pergamon Journals Ltd.

## A NOVEL WITHANOLIDE FROM *DATURA METEL*

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(Received 5 January 1987)

**Key Word Index**—*Datura metel*; Solanaceae; withanolide; Daturilin.

**Abstract**—A novel withanolide, daturilin, has been isolated from the alcohol soluble extract of the fresh leaves of *D. metel* and its structure established as 1-oxo-21,24S-epoxy-(20S,22S)-witha-2,5,25-trienolide through spectral studies including 2D NMR.

### INTRODUCTION

In an earlier communication the isolation and structure elucidation of a new tropane alkaloid datumetine [1] has been reported from the alcoholic extract of the leaves of *D. metel* L. Extended studies in the neutral constituents of the same plant material have led to the isolation and structure elucidation of a novel withanolide, daturilin (1). A few withanolides have been reported earlier from other species of *Datura* eg. *D. quercifolia*, [2–6], *D. ferox* [7], *D. stramonium* [8] and its variety *violaceae* [9], but this is the first instance of the isolation of a withanolide from *D. metel*. Further, although the C-21, C-24 epoxide ring has been reported in the triterpenes of tirucallane skeleton [10, 11], there is no precedence for an epoxide ring in a withanolide skeleton.

### RESULTS AND DISCUSSION

From the acid-insoluble fraction of the alcoholic extract of *D. metel* leaves, a colourless crystalline constituent daturilin (1) mp 206° has been obtained through solvent fractionation and preparative thin layer chromatography. The molecular ion  $[M]^+$  at  $m/z$  436 and high resolution MS corresponded to the molecular formula  $C_{28}H_{36}O_4$  and the prominent peaks at  $m/z$  269 and 268 were indicative of M-side chain fragment with and without transfer of hydrogen. The intensity and position of the UV band at 218 nm ( $\epsilon$  16318) indicated two  $\alpha,\beta$ -unsaturated carbonyl chromophores without extended conjugation. The IR spectrum exhibited bands at 1680 and 1720  $cm^{-1}$

for  $\alpha,\beta$ -unsaturated ketone and  $\alpha,\beta$ -unsaturated  $\delta$ -lactone groupings respectively [12], both characteristic of withanolides. IR and MS spectra did not show the presence of any hydroxyl group in 1, which was also confirmed through negative results of attempted acetylation. In the low field region of the  $^1H$  NMR spectrum of 1, signals related to five vinylic proton were observed. Two one-proton doublets of double doublets at  $\delta$  5.78 ( $J = 10.11, 3.24, 1.23$  Hz) and  $\delta$  6.69 ( $J = 10.11, 5.02, 2.49$  Hz) have been assigned to H-2 and H-3 respectively. H-6 appeared as a double triplet at  $\delta$  5.60 ( $J = 6.06, W_{1/2} = 1.71$  Hz) and the remaining two vinyl protons resonating at  $\delta$  5.99 and  $\delta$  6.56 as narrow doublets ( $J = 0.78$  Hz) were indicative of H-27a and H-27b of the exo double bond in the side

